

# Molecular dynamics simulation of the growth of carbon films obtained depositing C<sub>60</sub> fullerenes on silicon substrates



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## INTRODUCTION

Hard carbon materials obtained from polymerized or amorphized C<sub>60</sub> molecules have shown very interesting properties. Several experimental works have been performed in order to obtain hard carbon coatings from C<sub>60</sub> molecules accelerated at a defined deposition energy (E<sub>dep</sub>) [1,2]. From the theoretical point of view only the impact of individual molecules has been studied. We present here a theoretical simulation of the growth of films by sequential deposition of C<sub>60</sub> molecules on silicon and diamond substrates.

## COMPUTATIONAL MODEL

### Molecular dynamics

\* Semiempirical many-body Tersoff potential for C-C, C-Si and Si-Si interactions [3]

\* Berendsen's thermal bath [4]: T = 300 K; τ = 50Δt

### Substrates

**Diamond (100)** and **Silicon (100)**: 8 layers (2 fixed) of 50 atoms.  
Initial conditions: P=0 T=300 K (during 1000 fs)

### Deposition of C<sub>60</sub> molecules

Normal incidence. Initial random orientation.

Molecules vibrating at 300 K.

E<sub>dep</sub> from 10 to 1000 eV.

**Individual molecules**: studied during 3000 fs.

**Films**: ~ 30 molecules; 5000 fs between impacts.

## RESULTS

### Deposition of individual C<sub>60</sub> molecules

#### \* On Si (100) and diamond (100) substrates

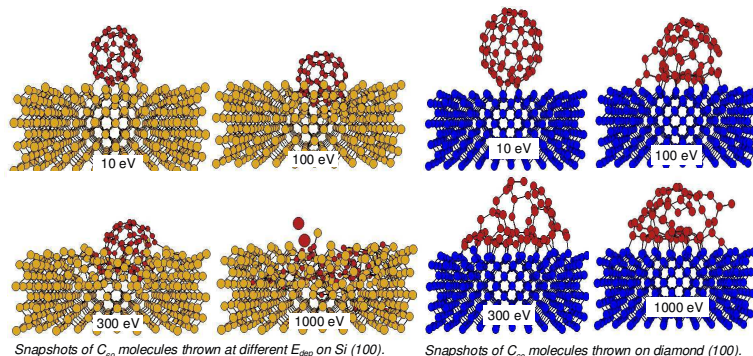
E<sub>dep</sub> ≤ 10 eV: the molecule either bounces away or remain weakly bonded to the substrate.

100 eV < E<sub>dep</sub> < 300 eV: increasing number of bonds and distortion of the fullerene cage with E<sub>dep</sub>.

E<sub>dep</sub> > 300 eV: the molecule is fragmented, some pieces bond to the substrate and others bounce away.

#### \* On diamond (111) substrates

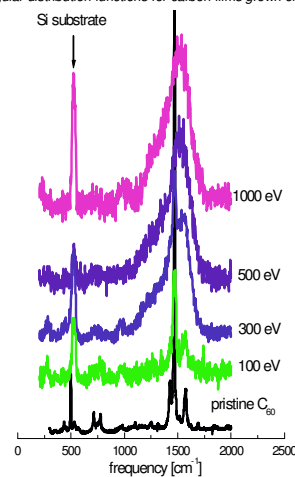
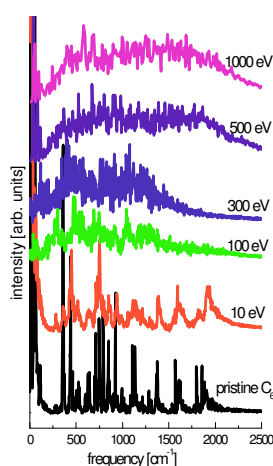
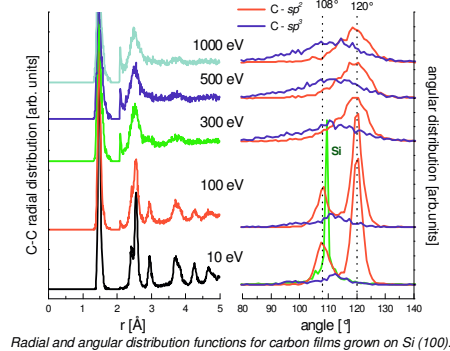
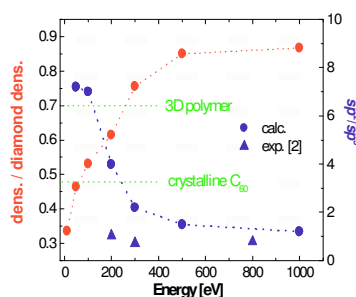
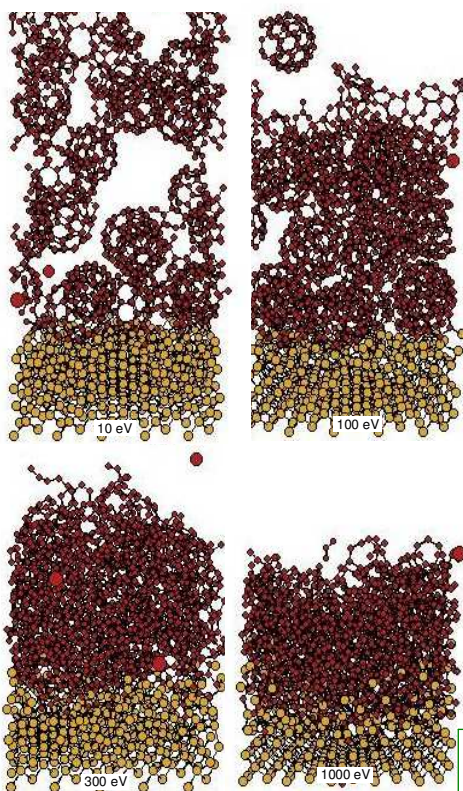
The threshold energy for fragmentation of fullerenes is significantly lower (~100 eV).



### Growth of carbon films by deposition of C<sub>60</sub> molecules

E<sub>dep</sub> < 300 eV: C<sub>60</sub> preserve the molecular identity, but some intermolecular covalent bonds begin to insinuate.

E<sub>dep</sub> > 300 eV: C<sub>60</sub> molecules are fragmented forming high density films. Density and substrate distortion increase with E<sub>dep</sub>.



## CONCLUSIONS

The penetration depth of C<sub>60</sub> is higher for Si than for diamond substrates, but in both cases is lower than the corresponding to C atoms [5]. This is in agreement with the experimentally observed poor adherence for C<sub>60</sub> on Si.

Molecular identity is preserved for E<sub>dep</sub> < 300 eV; for higher energies the films exhibit an increasing disorder until reaching the characteristics of amorphous carbon films resembling to those obtained from individual carbon atoms (the peaks corresponding to C<sub>60</sub> molecules are lost in radial and angular distribution functions; long range order is lost). This is in agreement with structural changes evidenced by experimental Raman spectra of films grown from C<sub>60</sub> ion beams [1].

The use of C<sub>60</sub> molecules as projectiles allows to grow dense amorphous carbon films in a wide range of energies and it is reasonable to assume that they would have interesting tribological properties, based on hardness measurements from similar films grown by other methods.

## REFERENCES

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